

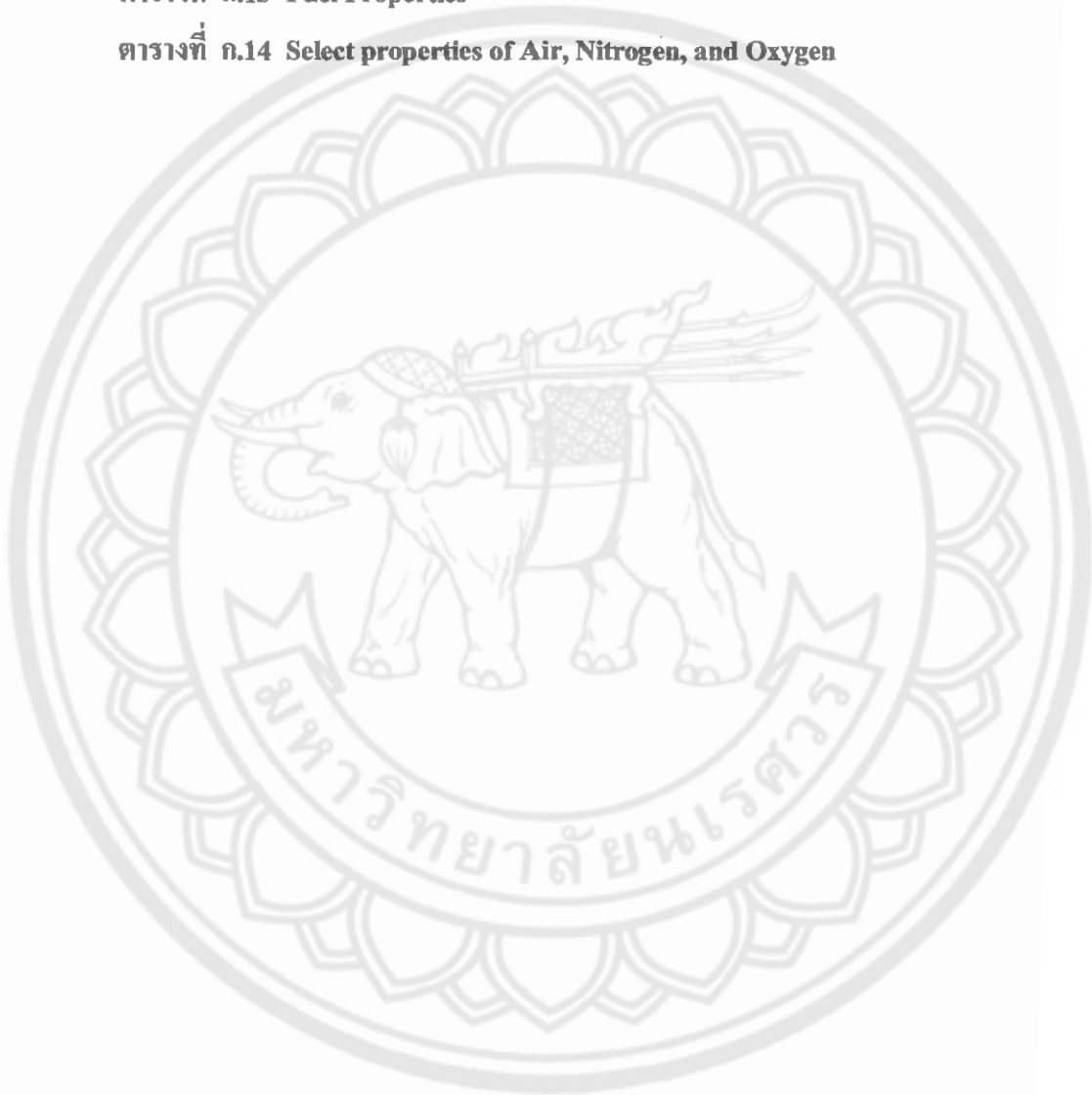


ภาคผนวก ก.

ตารางที่ ก.1 – ก.12 Selected thermodynamic properties of gases comprising
C-H-O-N system

ตารางที่ ก.13 Fuel Properties

ตารางที่ ก.14 Select properties of Air, Nitrogen, and Oxygen



Carbon dioxide (CO₂), MW = 44.011, enthalpy of formation @ 298 K
 (kJ/kmol) = -393.546

T(K)	$\bar{c}_p$ (kJ/kmol-K)	$(\bar{h}^o(T) - \bar{h}^o(298))$ (kJ/kmol)	$\bar{h}_f^o(T)$ (kJ/kmol)	$\bar{s}^o(T)$ (kJ/kmol-K)	$\bar{g}_f^o(T)$ (kJ/kmol)
200	32.387	-3.423	-393.483	199.876	-394.126
250	32.198	0	-393.546	213.736	-394.428
300	32.280	69	-393.547	213.966	-394.433
400	41.276	4,003	-393.617	225.257	-394.718
500	44.569	8,301	-393.712	234.833	-394.983
600	47.313	12,899	-393.844	243.209	-395.226
700	49.617	17,749	-394.013	250.680	-395.443
800	51.550	22,810	-394.213	257.436	-395.635
900	53.136	28,047	-394.433	263.603	-395.799
1,000	54.360	33,425	-394.659	269.268	-395.939
1,100	55.333	38,911	-394.875	274.495	-396.056
1,200	56.205	44,488	-395.083	279.348	-396.155
1,300	56.984	50,149	-395.287	283.878	-396.236
1,400	57.677	55,882	-395.488	288.127	-396.301
1,500	58.292	61,681	-395.691	292.128	-396.352
1,600	58.836	67,538	-395.897	295.908	-396.389
1,700	59.316	73,446	-396.110	299.489	-396.414
1,800	59.738	79,399	-396.332	302.892	-396.425
1,900	60.108	85,392	-396.564	306.132	-396.424
2,000	60.433	91,420	-396.808	309.223	-396.410
2,100	60.717	97,477	-397.065	312.179	-396.384
2,200	60.966	103,562	-397.338	315.009	-396.346
2,300	61.185	109,670	-397.626	317.724	-396.294
2,400	61.378	115,798	-397.931	320.333	-396.230
2,500	61.548	121,944	-398.253	322.842	-396.152
2,600	61.701	128,107	-398.594	325.259	-396.061
2,700	61.839	134,284	-398.952	327.590	-395.957
2,800	61.965	140,474	-399.329	329.841	-395.840
2,900	62.083	146,677	-399.725	332.018	-395.708
3,000	62.194	152,891	-400.140	334.124	-395.562
3,100	62.301	159,116	-400.573	336.165	-395.403
3,200	62.406	165,351	-401.025	338.145	-395.229
3,300	62.510	171,597	-401.495	340.067	-395.041
3,400	62.614	177,853	-401.983	341.935	-394.838
3,500	62.718	184,120	-402.489	343.751	-394.620
3,600	62.825	190,397	-403.013	345.519	-394.388
3,700	62.932	196,685	-403.553	347.242	-394.141
3,800	63.041	202,983	-404.110	348.922	-393.879
3,900	63.151	209,293	-404.684	350.561	-393.602
4,000	63.261	215,613	-405.273	353.161	-393.311
4,100	63.369	221,945	-405.878	355.725	-393.004
4,200	63.474	228,287	-406.499	358.253	-392.683
4,300	63.575	234,640	-407.135	359.748	-392.346
4,400	63.669	241,002	-407.785	358.210	-391.995
4,500	63.753	247,373	-408.451	359.642	-391.629
4,600	63.825	253,752	-409.132	361.044	-391.247
4,700	63.881	260,138	-409.828	362.417	-390.851

ตารางที่ 2 Hydrogen (H_2), MW = 2.016, enthalpy of formation @ 298 K (kJ/kmol) = 0

T (K)	$\bar{c}_p$ (kJ/kmol-K)	$(\bar{h}^o(T) - \bar{h}_f^o(298))$ (kJ/kmol)	$\bar{h}_f^o(T)$ (kJ/kmol)	$\bar{s}^o(T)$ (kJ/kmol-K)	$\bar{s}_f^o(T)$ (kJ/kmol-K)
200	28.522	-2.818	0	119.137	0
250	28.871	0	0	130.595	0
300	28.877	53	0	130.773	0
350	29.120	2,934	0	139.116	0
400	29.273	5,874	0	145.632	0
450	29.375	8,807	0	150.979	0
500	29.461	11,749	0	155.514	0
550	29.551	14,701	0	159.455	0
600	29.792	17,668	0	162.950	0
650	30.160	20,664	0	166.106	0
700	30.625	23,704	0	169.001	0
750	31.077	26,789	0	171.687	0
800	31.516	29,919	0	174.192	0
850	31.943	33,092	0	176.543	0
900	32.356	36,307	0	178.761	0
950	32.758	39,562	0	180.862	0
1,000	33.146	42,858	0	182.860	0
1,100	33.522	46,191	0	184.765	0
1,200	33.885	49,562	0	186.587	0
1,300	34.236	52,968	0	188.334	0
1,400	34.575	56,408	0	190.013	0
1,500	34.901	59,882	0	191.629	0
1,600	35.216	63,388	0	193.187	0
1,700	35.519	66,925	0	194.692	0
1,800	35.811	70,492	0	196.148	0
1,900	36.091	74,087	0	197.558	0
2,000	36.361	77,710	0	198.926	0
2,100	36.621	81,359	0	200.253	0
2,200	36.871	85,033	0	201.542	0
2,300	37.112	88,733	0	202.796	0
2,400	37.343	92,455	0	204.017	0
2,500	37.566	96,201	0	205.206	0
2,600	37.781	99,968	0	206.365	0
2,700	37.989	103,757	0	207.496	0
2,800	38.190	107,566	0	208.600	0
2,900	38.385	111,395	0	209.679	0
3,000	38.574	115,243	0	210.733	0
3,100	38.759	119,109	0	211.764	0
3,200	38.939	122,994	0	212.774	0
3,300	39.116	126,897	0	213.762	0
3,400	39.291	130,817	0	214.730	0
3,500	39.464	134,755	0	215.679	0
3,600	39.636	138,710	0	216.609	0
3,700	39.808	142,682	0	217.522	0
3,800	39.981	146,672	0	218.419	0
3,900	40.156	150,679	0	219.300	0
4,000	40.334	154,703	0	220.165	0

таблица 0.3 Carbon monoxide (CO), MW = 28.010, enthalpy of formation @ 298 K
 (kJ/kmol) = -110.541

T(K)	c_p (kJ/kmol-K)	$(\bar{h}^o(T) - \bar{h}_f^o(298))$ (kJ/kmol)	$\bar{h}_f^o(T)$ (kJ/kmol)	$s^o(T)$ (kJ/kmol-K)	$\bar{g}_f^o(T)$ (kJ/kmol)
200	28.687	-2.835	-111.308	186.018	-128.532
250	29.072	0	-110.541	197.548	-137.163
300	29.078	54	-110.530	197.728	-137.328
400	29.433	3.979	-110.121	206.141	-146.332
500	29.857	5.943	-110.017	212.752	-155.403
600	30.407	8.955	-110.156	218.247	-164.470
700	31.089	12.029	-110.477	222.979	-173.499
800	31.860	15.176	-110.924	227.180	-182.475
900	32.629	18.401	-111.450	230.978	-191.386
1,000	33.255	21.697	-112.022	234.450	-200.238
1,100	33.725	25.046	-112.619	237.602	-209.050
1,200	34.148	28.440	-113.240	240.595	-217.768
1,300	34.530	31.874	-113.881	243.344	-226.483
1,400	34.872	35.345	-114.543	245.915	-235.087
1,500	35.178	38.847	-115.225	248.332	-243.674
1,600	35.451	42.379	-115.925	250.611	-252.214
1,700	35.694	45.937	-116.644	252.768	-260.711
1,800	35.910	49.517	-117.380	254.814	-269.164
1,900	36.101	53.118	-118.132	256.761	-277.576
2,000	36.271	56.737	-118.902	258.617	-285.948
2,100	36.421	60.371	-119.687	260.391	-294.281
2,200	36.553	64.020	-120.488	262.088	-302.576
2,300	36.670	67.682	-121.305	263.715	-310.835
2,400	36.774	71.354	-122.137	265.278	-319.057
2,500	36.867	75.036	-122.984	266.781	-327.245
2,600	36.950	78.727	-123.847	268.229	-335.399
2,700	37.025	82.426	-124.724	269.625	-343.519
2,800	37.093	86.132	-125.616	270.973	-351.606
2,900	37.155	89.844	-126.523	272.275	-359.661
3,000	37.213	93.562	-127.446	273.536	-367.684
3,100	37.268	97.287	-128.383	274.757	-375.677
3,200	37.321	101.016	-129.335	275.941	-383.639
3,300	37.372	104.751	-130.303	277.090	-391.571
3,400	37.422	108.490	-131.285	278.207	-399.474
3,500	37.471	112.235	-132.283	279.292	-407.347
3,600	37.521	115.985	-133.295	280.349	-415.192
3,700	37.570	119.739	-134.323	281.377	-423.008
3,800	37.619	123.499	-135.366	282.380	-430.796
3,900	37.667	127.263	-136.424	283.358	-438.557
4,000	37.716	131.032	-137.497	284.312	-446.291
4,100	37.764	134.806	-138.585	285.244	-453.997
4,200	37.810	138.585	-139.687	286.154	-461.677
4,300	37.855	142.368	-140.804	287.045	-469.330
4,400	37.897	146.156	-141.935	287.915	-476.957
4,500	37.936	149.948	-143.079	288.768	-484.558
4,600	37.970	153.743	-144.236	289.602	-492.134
4,700	37.998	157.541	-145.407	290.419	-499.684

ตารางที่ 4 Hydroxyl (OH), MW = 17.007, enthalpy of formation @ 298 K (kJ/kmol)
= 33.985

T(K)	$\bar{c}_p$ (kJ/kmol-K)	$(\bar{h}^o(T) - \bar{h}^o(298))$ (kJ/kmol)	$\bar{h}_f^o(T)$ (kJ/kmol)	$\bar{s}^o(T)$ (kJ/kmol-K)	$\bar{g}_f^o(T)$ (kJ/kmol)
200	30.140	-2.948	38.864	171.607	35.808
250	29.932	0	38.985	183.604	34.279
300	29.928	55	38.987	183.789	34.250
400	29.718	3.037	39.030	192.369	32.662
500	29.570	6.001	39.000	198.983	31.072
600	29.527	8.955	38.909	204.369	29.494
700	29.615	11.911	38.770	208.925	27.935
800	29.844	14.883	38.599	212.893	26.399
900	30.208	17.884	38.410	216.428	24.885
1000	30.682	20.928	38.220	219.635	23.392
1100	31.186	24.022	38.039	222.583	21.918
1200	31.662	27.164	37.867	225.317	20.460
1300	32.114	30.353	37.704	227.869	19.017
1400	32.540	33.586	37.548	230.265	17.585
1500	32.943	36.860	37.397	232.524	16.164
1600	33.323	40.174	37.252	234.662	14.753
1700	33.682	43.524	37.109	236.693	13.352
1800	34.019	46.910	36.969	238.628	11.958
1900	34.337	50.328	36.831	240.476	10.573
2000	34.635	53.776	36.693	242.245	9.194
2100	34.915	57.254	36.555	243.942	7.823
2200	35.178	60.759	36.416	245.572	6.458
2300	35.425	64.289	36.276	247.141	5.099
2400	35.656	67.843	36.133	248.654	3.746
2500	35.872	71.420	35.986	250.114	2.400
2600	36.074	75.017	35.836	251.525	1.060
2700	36.263	78.634	35.682	252.890	-0.275
2800	36.439	82.269	35.524	254.212	-1.604
2900	36.604	85.922	35.360	255.493	-2.927
3000	36.759	89.590	35.191	256.737	-4.245
3100	36.903	93.273	35.016	257.943	-5.556
3200	37.039	96.970	34.835	259.118	-6.862
3300	37.166	100.681	34.648	260.260	-8.162
3400	37.285	104.403	34.454	261.371	-9.457
3500	37.398	108.137	34.253	262.454	-10.745
3600	37.504	111.882	34.046	263.509	-12.028
3700	37.605	115.638	33.831	264.538	-13.305
3800	37.701	119.403	33.610	265.542	-14.576
3900	37.793	123.178	33.381	266.522	-15.841
4000	37.882	126.962	33.146	267.480	-17.100
4100	37.968	130.754	32.903	268.417	-18.353
4200	38.052	134.555	32.654	269.333	-19.600
4300	38.135	138.365	32.397	270.229	-20.841
4400	38.217	142.182	32.134	271.107	-22.076
4500	38.300	146.008	31.864	271.967	-23.306
4600	38.382	149.842	31.588	272.809	-24.528
4700	38.466	153.685	31.305	273.636	-25.745

附表 A.5 Nitrogen (N_2), MW = 28.013, enthalpy of formation @ 298 K (kJ/kmol) = 0

T(K)	c_p (kJ/kmol-K)	$(\hat{h}^o(T) - \hat{h}^o(298))$ (kJ/kmol)	$\hat{h}_f^o(T)$ (kJ/kmol)	$s^o(T)$ (kJ/kmol-K)	$\hat{s}_f^o(T)$ (kJ/kmol)
200	28.793	2.841	0	179.959	0
298	29.071	0	0	191.511	0
300	29.075	54	0	191.691	0
400	29.319	2.973	0	200.088	0
500	29.636	5.920	0	206.662	0
600	30.086	8.905	0	212.103	0
700	30.684	11.942	0	216.784	0
800	31.394	15.046	0	220.927	0
900	32.131	18.232	0	224.667	0
1000	32.762	21.468	0	228.087	0
1100	33.288	24.770	0	231.233	0
1200	33.707	28.118	0	234.146	0
1300	34.117	31.510	0	236.861	0
1400	34.477	34.939	0	239.402	0
1500	34.803	38.404	0	241.792	0
1600	35.099	41.899	0	244.048	0
1700	35.361	45.423	0	246.184	0
1800	35.595	48.971	0	248.212	0
1900	35.803	52.541	0	250.142	0
2000	35.988	56.130	0	251.983	0
2100	36.152	59.738	0	253.743	0
2200	36.298	63.360	0	255.429	0
2300	36.428	66.997	0	257.045	0
2400	36.543	70.645	0	258.598	0
2500	36.645	74.305	0	260.092	0
2600	36.737	77.974	0	261.531	0
2700	36.820	81.652	0	262.919	0
2800	36.895	85.338	0	264.259	0
2900	36.964	89.031	0	265.555	0
3000	37.028	92.730	0	266.810	0
3100	37.088	96.436	0	268.025	0
3200	37.144	100.148	0	269.203	0
3300	37.198	103.865	0	270.347	0
3400	37.251	107.587	0	271.458	0
3500	37.302	111.315	0	272.539	0
3600	37.352	115.048	0	273.590	0
3700	37.402	118.786	0	274.614	0
3800	37.452	122.528	0	275.612	0
3900	37.501	126.276	0	276.586	0
4000	37.549	130.028	0	277.536	0
4100	37.597	133.786	0	278.464	0
4200	37.643	137.548	0	279.370	0
4300	37.688	141.314	0	280.257	0
4400	37.730	145.085	0	281.123	0
4500	37.768	148.860	0	281.972	0
4600	37.803	152.639	0	282.802	0
4700	37.832	156.420	0	283.616	0

ตาราง 6.6 Water (H_2O , MW = 18.016, enthalpy of formation @ 298 K (kJ/kmol) = -241.845, enthalpy of vaporization (kJ/kmol) = 44,010

T(K)	c_p (kJ/kmol-K)	$(H(T) - H_f(298))$ (kJ/kmol)	$H_f(T)$ (kJ/kmol)	$s^*(T)$ (kJ/kmol-K)	$H_g(T)$ (kJ/kmol)
200	32.255	-3.227	-240.838	135.602	-237.779
250	33.448	0	-241.845	188.715	-228.608
300	34.468	62	-241.865	198.922	-228.526
350	35.437	1453	-242.858	198.686	-223.929
400	36.357	6.947	-243.822	206.467	-219.085
450	37.225	10.528	-244.753	212.992	-214.049
500	37.364	14.209	-245.638	218.665	-208.861
550	38.587	18.005	-246.461	223.733	-203.550
600	39.930	21.930	-247.209	228.354	-198.141
650	41.515	25.993	-247.879	232.633	-192.652
700	42.658	30.191	-248.475	236.634	-187.100
750	43.574	34.518	-249.005	240.397	-181.497
800	44.057	38.963	-249.477	243.955	-175.852
850	46.162	43.520	-249.895	247.332	-170.172
900	47.115	48.181	-250.267	250.547	-164.464
950	48.028	52.939	-250.597	253.617	-158.733
1000	48.901	57.786	-250.890	256.556	-152.983
1050	49.705	62.715	-251.151	259.374	-147.216
1100	50.451	67.725	-251.384	262.081	-141.435
1150	51.147	72.805	-251.594	264.687	-135.643
1200	51.784	77.952	-251.783	267.198	-129.841
1250	52.378	83.160	-251.953	269.621	-124.030
1300	52.927	88.426	-252.113	271.961	-118.211
1350	53.435	93.744	-252.261	274.225	-112.386
1400	53.905	99.112	-252.399	276.416	-106.555
1450	54.320	104.524	-252.532	278.539	-100.719
1500	54.702	109.979	-252.659	280.597	-94.878
1550	55.115	115.472	-252.785	282.595	-89.031
1600	55.459	121.001	-252.909	284.535	-83.181
1650	55.779	126.563	-253.034	286.420	-77.326
1700	56.076	132.156	-253.161	288.254	-71.467
1750	56.353	137.777	-253.290	290.039	-65.604
1800	56.610	143.426	-253.423	291.777	-59.737
1850	56.851	149.099	-253.561	293.471	-53.865
1900	57.076	154.795	-253.704	295.122	-47.990
1950	57.288	160.514	-253.852	296.733	-42.110
2000	57.488	166.252	-254.007	298.305	-36.226
2050	57.676	172.011	-254.169	299.841	-30.338
2100	57.856	177.787	-254.338	301.341	-24.446
2150	58.026	183.583	-254.515	302.808	-18.549
2200	58.190	189.392	-254.699	304.243	-12.648
2250	58.346	195.219	-254.892	305.647	-6.742
2300	58.496	201.061	-255.093	307.022	-0.831
2350	58.641	206.915	-255.303	308.368	5.085
2400	58.781	212.790	-255.522	309.688	11.005
2450	58.916	218.674	-255.751	310.981	16.930
2500	59.047	224.573	-255.990	312.250	22.861

Н1314 п. 7 Nitrogen atom (N), MW = 14.007, enthalpy of formation @ 298 K (kJ/kmol)
= 472.629

T(K)	c_p (kJ/kmol-K)	$(h^o(T) - h^o_{f,298})$ (kJ/kmol)	$\tilde{h}_f^o(T)$ (kJ/kmol)	$\tilde{s}^o(T)$ (kJ/kmol-K)	$\tilde{g}_f^o(T)$ (kJ/kmol)
200	20.790	-2.040	472.008	144.889	461.026
250	20.786	0	472.629	153.189	455.504
300	20.786	38	472.640	153.317	455.398
400	20.786	2.117	473.258	159.297	449.557
500	20.786	4.196	473.864	163.933	443.562
600	20.786	6.274	474.450	167.725	437.446
700	20.786	8.353	475.010	170.929	431.234
800	20.786	10.431	475.537	173.705	424.944
900	20.786	12.510	476.027	176.153	418.590
1 000	20.786	14.589	476.483	178.343	412.183
1 100	20.792	16.668	476.911	180.325	405.733
1 200	20.795	18.747	477.316	182.134	399.243
1 300	20.795	20.826	477.700	183.798	392.721
1 400	20.793	22.906	478.064	185.339	386.171
1 500	20.790	24.985	478.411	186.774	379.595
1 600	20.786	27.064	478.742	188.115	372.996
1 700	20.782	29.142	479.059	189.375	366.377
1 800	20.779	31.220	479.363	190.563	359.740
1 900	20.777	33.298	479.656	191.687	353.086
2 000	20.776	35.376	479.939	192.752	346.417
2 100	20.778	37.453	480.213	193.766	339.735
2 200	20.783	39.531	480.479	194.733	333.039
2 300	20.791	41.610	480.740	195.657	326.331
2 400	20.802	43.690	480.995	196.542	319.612
2 500	20.818	45.771	481.246	197.391	312.883
2 600	20.838	47.853	481.494	198.208	306.143
2 700	20.864	49.938	481.740	198.995	299.394
2 800	20.895	52.026	481.985	199.754	292.636
2 900	20.931	54.118	482.230	200.488	285.870
3 000	20.974	56.213	482.476	201.199	279.094
3 100	21.024	58.313	482.723	201.887	272.311
3 200	21.080	60.418	482.972	202.555	265.519
3 300	21.143	62.529	483.224	203.205	258.720
3 400	21.214	64.647	483.481	203.837	251.913
3 500	21.292	66.772	483.742	204.453	245.099
3 600	21.378	68.905	484.009	205.054	238.276
3 700	21.472	71.048	484.283	205.641	231.447
3 800	21.575	73.200	484.564	206.213	224.610
3 900	21.686	75.363	484.853	206.777	217.765
4 000	21.805	77.537	485.151	207.328	210.913
4 100	21.934	79.724	485.459	207.868	204.053
4 200	22.071	81.924	485.779	208.398	197.186
4 300	22.217	84.139	486.110	208.919	190.310
4 400	22.372	86.368	486.453	209.431	183.427
4 500	22.536	88.613	486.811	209.936	176.536
4 600	22.709	90.875	487.184	210.433	169.637
4 700	22.891	93.155	487.573	210.923	162.730

81334 n.8
= 90,297

Nitric oxide (NO), MW = 30.006, enthalpy of formation @ 298 K (kJ/kmol)

T(K)	ε_f (kJ/kmol-K)	$(\bar{h}^o(T) - \bar{h}_f^o(298))$ (kJ/kmol)	$\bar{h}_f^o(T)$ (kJ/kmol)	$\bar{s}^o(T)$ (kJ/kmol-K)	$\bar{g}_f^o(T)$ (kJ/kmol)
200	29.374	-2.901	90.234	198.856	87.811
298	29.728	0	90.297	210.652	86.607
300	29.735	55	90.298	210.836	86.584
400	30.103	3.046	90.341	219.439	85.340
500	30.570	6.079	90.367	226.204	84.086
600	31.174	9.165	90.382	231.829	82.828
700	31.908	12.318	90.393	236.688	81.563
800	32.715	15.549	90.405	241.001	80.307
900	33.489	18.860	90.421	244.900	79.043
1 000	34.076	22.241	90.443	248.462	77.778
1 100	34.483	25.669	90.465	251.729	76.510
1 200	34.850	29.136	90.486	254.745	75.241
1 300	35.180	32.638	90.505	257.548	73.970
1 400	35.474	36.171	90.520	260.166	72.697
1 500	35.737	39.732	90.532	262.623	71.423
1 600	35.972	43.317	90.538	264.937	70.149
1 700	36.180	46.925	90.539	267.124	68.875
1 800	36.364	50.552	90.534	269.197	67.601
1 900	36.527	54.197	90.523	271.168	66.327
2 000	36.671	57.857	90.505	273.045	65.054
2 100	36.797	61.531	90.479	274.838	63.782
2 200	36.909	65.216	90.447	276.552	62.511
2 300	37.008	68.912	90.406	278.195	61.243
2 400	37.095	72.617	90.358	279.772	59.976
2 500	37.173	76.331	90.303	281.288	58.711
2 600	37.242	80.052	90.239	282.747	57.448
2 700	37.303	83.779	90.168	284.151	56.188
2 800	37.362	87.513	90.089	285.512	54.931
2 900	37.415	91.251	90.003	286.824	53.677
3 000	37.464	94.995	89.909	288.093	52.426
3 100	37.511	98.744	89.809	289.322	51.178
3 200	37.556	102.498	89.701	290.514	49.934
3 300	37.600	106.255	89.586	291.670	48.693
3 400	37.643	110.018	89.465	292.793	47.456
3 500	37.686	113.784	89.337	293.885	46.222
3 600	37.729	117.555	89.203	294.947	44.992
3 700	37.771	121.330	89.063	295.981	43.766
3 800	37.815	125.109	88.918	296.989	42.543
3 900	37.858	128.893	88.767	297.972	41.325
4 000	37.900	132.680	88.611	298.931	40.110
4 100	37.943	136.473	88.449	299.867	38.900
4 200	37.984	140.269	88.283	300.782	37.693
4 300	38.023	144.069	88.112	301.677	36.491
4 400	38.060	147.873	87.936	302.551	35.292
4 500	38.093	151.681	87.755	303.407	34.098
4 600	38.122	155.492	87.569	304.244	32.908
4 700	38.146	159.305	87.379	305.064	31.721

Table B.9 Nitrogen dioxide (NO_2), MW = 46.006, enthalpy of formation @ 298 K
 (kJ/kmol) = 33.098

T(K)	$\bar{c}_p$ (kJ/kmol-K)	$(\bar{h}^\circ(T) - \bar{h}_f^\circ(298))$ (kJ/kmol)	$\bar{h}_f^\circ(T)$ (kJ/kmol)	$\bar{s}^\circ(T)$ (kJ/kmol-K)	$\bar{g}_f^\circ(T)$ (kJ/kmol)
200	32.936	-3.432	33.961	226.016	45.453
250	36.881	0	33.098	239.925	51.291
300	36.949	68	33.083	240.153	51.403
400	40.331	3.937	32.521	251.259	57.602
500	44.227	8.118	32.173	260.578	63.916
600	45.737	12.569	31.974	268.686	70.285
700	47.913	17.255	31.885	275.904	76.679
800	49.762	22.141	31.880	282.427	83.079
900	51.243	27.195	31.938	288.377	89.476
1,000	52.271	32.375	32.035	293.834	95.864
1,100	52.980	37.638	32.146	298.850	102.242
1,200	53.625	42.970	32.267	303.489	108.609
1,300	54.186	48.361	32.392	307.801	114.966
1,400	54.679	53.805	32.519	311.838	121.313
1,500	55.109	59.295	32.643	315.625	127.651
1,600	55.483	64.825	32.762	319.194	133.981
1,700	55.805	70.390	32.875	322.568	140.303
1,800	56.082	75.984	32.973	325.765	146.620
1,900	56.318	81.605	33.061	328.804	152.931
2,000	56.517	87.247	33.134	331.698	159.238
2,100	56.685	92.905	33.192	334.460	165.542
2,200	56.826	98.583	33.233	337.100	171.843
2,300	56.943	104.271	33.256	339.629	178.143
2,400	57.040	109.971	33.262	342.054	184.442
2,500	57.121	115.679	33.248	344.384	190.742
2,600	57.188	121.304	33.216	346.626	197.042
2,700	57.244	127.116	33.165	348.785	203.344
2,800	57.291	132.843	33.093	350.868	209.648
2,900	57.333	138.574	33.005	352.879	215.955
3,000	57.371	144.309	32.900	354.824	222.265
3,100	57.406	150.048	32.776	356.705	228.579
3,200	57.440	155.791	32.634	358.529	234.898
3,300	57.474	161.536	32.476	360.297	241.221
3,400	57.509	167.285	32.302	362.013	247.549
3,500	57.546	173.038	32.113	363.680	253.883
3,600	57.584	178.795	31.908	365.302	260.222
3,700	57.624	184.555	31.689	366.880	266.567
3,800	57.665	190.319	31.456	368.418	272.918
3,900	57.708	196.088	31.210	369.916	279.276
4,000	57.750	201.861	30.951	371.378	285.639
4,100	57.792	207.638	30.678	372.804	292.010
4,200	57.831	213.419	30.393	374.197	298.387
4,300	57.866	219.204	30.093	375.559	304.772
4,400	57.895	224.992	29.783	376.889	311.163
4,500	57.915	230.783	29.457	378.190	317.562
4,600	57.925	236.575	29.117	379.464	323.968
4,700	57.922	242.367	28.761	380.709	330.381

A.13.14 A.10

Oxygen (O_2), MW = 31.999, enthalpy of formation @ 298 K (kJ/kmol) = 0

T (K)	$\bar{c}_p$ (kJ/kmol-K)	$(\bar{h}^*(T) - \bar{h}_f^*(298))$ (kJ/kmol)	$\bar{h}_f^*(T)$ (kJ/kmol)	$\bar{s}^*(T)$ (kJ/kmol-K)	$\bar{g}_f^*(T)$ (kJ/kmol)
200	28.473	-2.836	0	193.518	0
298	29.315	0	0	205.043	0
300	29.331	54	0	205.224	0
400	30.210	3.031	0	213.782	0
500	31.114	6.097	0	220.620	0
600	32.030	9.254	0	226.374	0
700	32.927	12.503	0	231.379	0
800	33.757	15.838	0	235.831	0
900	34.454	19.250	0	239.849	0
1,000	34.936	22.721	0	243.507	0
1,100	35.270	26.232	0	246.852	0
1,200	35.593	29.775	0	249.935	0
1,300	35.903	33.350	0	252.796	0
1,400	36.202	36.953	0	255.468	0
1,500	36.490	40.590	0	257.976	0
1,600	36.768	44.253	0	260.339	0
1,700	37.036	47.943	0	262.577	0
1,800	37.296	51.660	0	264.701	0
1,900	37.546	55.402	0	266.724	0
2,000	37.788	59.169	0	268.656	0
2,100	38.023	62.959	0	270.506	0
2,200	38.250	66.773	0	272.280	0
2,300	38.470	70.609	0	273.985	0
2,400	38.684	74.467	0	275.627	0
2,500	38.891	78.346	0	277.210	0
2,600	39.093	82.245	0	278.739	0
2,700	39.289	86.164	0	280.218	0
2,800	39.480	90.103	0	281.651	0
2,900	39.665	94.060	0	283.039	0
3,000	39.846	98.036	0	284.387	0
3,100	40.023	102.029	0	285.697	0
3,200	40.195	106.040	0	286.970	0
3,300	40.362	110.068	0	288.209	0
3,400	40.526	114.112	0	289.417	0
3,500	40.686	118.173	0	290.594	0
3,600	40.842	122.249	0	291.742	0
3,700	40.994	126.341	0	292.863	0
3,800	41.143	130.448	0	293.959	0
3,900	41.287	134.570	0	295.029	0
4,000	41.429	138.705	0	296.076	0
4,100	41.566	142.855	0	297.101	0
4,200	41.700	147.019	0	298.104	0
4,300	41.830	151.195	0	299.087	0
4,400	41.957	155.384	0	300.050	0
4,500	42.079	159.586	0	300.994	0
4,600	42.197	163.800	0	301.921	0
4,700	42.312	168.026	0	302.829	0

таблица 11
= 249.197

Oxygen atom (O), MW = 16.000, enthalpy of formation @ 298 K (kJ/kmol)

T(K)	$\bar{c}_p$ (kJ/kmol-K)	$(\bar{h}^o(T) - \bar{h}_f^o(298))$ (kJ/kmol)	$\bar{h}_f^o(T)$ (kJ/kmol)	$\bar{s}^o(T)$ (kJ/kmol-K)	$\bar{g}_f^o(T)$ (kJ/kmol)
200	22.377	-2.176	248.439	152.085	237.374
250	21.899	0	249.197	160.945	231.778
300	21.890	41	249.211	161.080	231.670
400	21.800	2.209	249.890	167.320	225.719
500	21.256	4.345	250.494	172.089	219.605
600	21.113	6.463	251.033	175.951	213.375
700	21.033	8.570	251.516	179.199	207.060
800	20.986	10.671	251.949	182.004	200.679
900	20.952	12.768	252.340	184.474	194.246
1000	20.915	14.861	252.698	186.679	187.772
1100	20.898	16.952	253.033	188.672	181.263
1200	20.882	19.041	253.350	190.490	174.724
1300	20.867	21.128	253.650	192.160	168.159
1400	20.854	23.214	253.934	193.706	161.572
1500	20.843	25.299	254.201	195.145	154.966
1600	20.834	27.383	254.454	196.490	148.342
1700	20.827	29.466	254.692	197.753	141.702
1800	20.822	31.548	254.916	198.943	135.049
1900	20.820	33.630	255.127	200.069	128.384
2000	20.819	35.712	255.325	201.136	121.709
2100	20.821	37.794	255.512	202.152	115.023
2200	20.825	39.877	255.687	203.121	108.329
2300	20.831	41.959	255.852	204.047	101.627
2400	20.840	44.043	256.007	204.933	94.918
2500	20.851	46.127	256.152	205.784	88.203
2600	20.865	48.213	256.288	206.602	81.483
2700	20.881	50.300	256.416	207.390	74.757
2800	20.899	52.389	256.535	208.150	68.037
2900	20.920	54.480	256.648	208.884	61.292
3000	20.944	56.574	256.753	209.593	54.534
3100	20.970	58.669	256.852	210.280	47.812
3200	20.998	60.768	256.945	210.947	41.068
3300	21.028	62.869	257.032	211.593	34.320
3400	21.061	64.973	257.114	212.221	27.570
3500	21.095	67.081	257.192	212.832	20.818
3600	21.132	69.192	257.265	213.427	14.063
3700	21.171	71.308	257.334	214.007	7.307
3800	21.212	73.427	257.400	214.572	548
3900	21.254	75.550	257.462	215.123	-6.312
4000	21.299	77.678	257.522	215.662	-12.974
4100	21.345	79.810	257.579	216.189	-19.737
4200	21.392	81.947	257.635	216.703	-26.501
4300	21.441	84.088	257.688	217.207	-33.267
4400	21.490	86.235	257.740	217.701	-40.034
4500	21.540	88.386	257.790	218.184	-46.802
4600	21.591	90.543	257.840	218.658	-53.571
4700	21.646	92.705	257.889	219.123	-60.342

A1914.N.13

Selected properties of air at 1 atm^a

T (K)	ρ (kg/m ³)	c_p (kJ/kg·K)	$\mu \cdot 10^7$ (N·s/m ²)	$\nu \cdot 10^6$ (m ² /s)	$k \cdot 10^5$ (W/m·K)	$\alpha \cdot 10^6$ (m ² /s)	Pr
100	1.2562	1.032	71.1	2.00	9.34	2.54	0.786
150	1.2364	1.012	103.4	4.426	13.8	5.84	0.758
200	1.2158	1.007	132.5	7.590	18.1	10.3	0.733
250	1.1947	1.006	159.6	11.44	22.3	15.9	0.720
300	1.1614	1.007	184.6	15.89	26.3	22.5	0.707
350	0.9950	1.009	208.2	20.92	30.0	29.9	0.700
400	0.8711	1.014	230.1	26.41	33.8	38.3	0.699
450	0.7740	1.021	250.7	32.39	37.3	47.2	0.686
500	0.6964	1.030	270.1	38.79	40.7	56.7	0.684
550	0.6329	1.040	288.4	45.57	43.9	66.7	0.683
600	0.5804	1.051	305.8	52.69	46.9	76.9	0.685
650	0.5356	1.063	322.5	60.21	49.7	87.3	0.690
700	0.4975	1.075	338.8	68.10	52.4	98.0	0.695
750	0.4643	1.087	354.6	76.37	54.9	109	0.702
800	0.4354	1.099	369.8	84.93	57.3	120	0.709
850	0.4097	1.110	384.3	93.80	59.6	131	0.716
900	0.3868	1.121	398.1	102.9	62.0	143	0.720
950	0.3666	1.131	411.3	112.2	64.3	155	0.723
1,000	0.3482	1.141	424.4	121.9	66.7	168	0.726
1,100	0.3166	1.159	449.0	141.8	71.5	195	0.728
1,200	0.2902	1.175	473.0	162.9	76.3	224	0.728
1,300	0.2679	1.189	496.0	185.1	82	238	0.719
1,400	0.2488	1.207	530	213	91	303	0.703
1,500	0.2322	1.230	557	240	100	370	0.685
1,600	0.2177	1.248	584	268	106	390	0.688

A13.9.12 Selected properties of hydrocarbon fuels: enthalpy of formation,^a Gibbs function of formation,^b entropy,^c and higher and lower heating values^d at 298.15 K and 1 atm; boiling points^e and critical heat of vaporization^f at 1 atm; constant-pressure adiabatic flame temperature at 1 atm;^g liquid density^h

Formula	Fuel	M (kg/kmol)	H_f° (kJ/kmol)	H_g° (kJ/kmol)	U_f° (kJ/kmol-K)	U_g° (kJ/kmol-K)	$H_{f,298}^\circ$ (kJ/kg)	$H_{f,298}^\circ$ (kJ/kg)	Boiling pt. (°C)	ρ_l (kg/m ³)	T_{ad}^h (K)	ρ_{ad}^h (kg/m ³)
CH_4	Methane	16.043	74,831	50,794	186.188	55.528	4600.6	48.225	164	429	2,226	460
C_2H_2	Acetylene	26.038	226,748	70,700	200.819	39,923	38.225	38.225	84	—	2,830	—
C_2H_4	Ethylene	28.054	52,283	68,134	219.827	50.333	47.161	47.161	103.7	—	2,160	—
C_2H_6	Ethane	30.069	84,667	32,886	229.492	51.901	47.489	47.489	88.6	488	2,259	470
C_3H_6	Propene	42.080	20,414	62,718	266.979	38.936	45.784	45.784	47.4	412	2,334	514
C_3H_8	Propane	44.096	103,847	21,489	269.910	50.368	46.357	46.357	42.1	425	2,267	540
C_4H_8	1-Butene	56.107	1,172	32,836	307.440	48.471	45.319	45.319	63	391	2,322	595
C_4H_{10}	<i>n</i> -Butane	58.123	124,733	14,707	310.034	49.546	45.742	45.742	-0.5	386	2,220	599
C_5H_{10}	1-Pentene	70.134	20,920	79,605	347.607	48.152	45.006	45.006	30	358	2,314	641
C_5H_{12}	<i>n</i> -Pentane	72.150	146,440	8,201	348.402	49.032	45.355	45.355	36.1	358	2,272	650
C_6H_6	Benzene	78.113	82,927	129,658	269.199	42.277	40.579	40.579	80.1	393	2,342	679
C_6H_{12}	1-Hexene	84.161	41,673	87,027	385.974	47.955	44.803	44.803	63.4	335	2,308	673
C_6H_{14}	<i>n</i> -Hexane	86.177	167,193	209	386.811	48.696	45.105	45.105	69	335	2,271	679
C_7H_{14}	1-Heptene	98.188	62,132	95,563	424.383	47.817	44.665	44.665	93.6	—	2,305	—
C_7H_{16}	<i>n</i> -Heptane	100.203	187,820	8,145	425.262	48.456	44.926	44.926	98.4	310	2,274	664
C_8H_{16}	1-Octene	112.214	82,927	104,140	462.792	47.712	44.560	44.560	121.3	—	2,302	—
C_8H_{18}	<i>n</i> -Octane	114.230	208,447	17,322	463.673	48.275	44.791	44.791	125.7	304	2,276	703
C_9H_{18}	1-Nonene	126.241	103,512	112,717	501.243	47.631	44.478	44.478	—	—	2,300	—
C_9H_{20}	<i>n</i> -Nonane	128.257	229,032	25,857	502.080	48.134	44.686	44.686	150.8	295	2,276	718
$C_{10}H_{18}$	1-Decene	140.268	124,139	121,394	539.652	47.565	44.413	44.413	170.6	—	2,295	—
$C_{10}H_{22}$	<i>n</i> -Decane	142.284	249,659	34,134	540.531	48.020	44.602	44.602	174.1	277	2,277	740
$C_{11}H_{22}$	1-Undecene	154.295	144,766	129,850	578.061	47.512	44.360	44.360	—	—	2,295	—
$C_{11}H_{24}$	<i>n</i> -Undecane	156.311	270,286	43,012	578.940	47.926	44.532	44.532	193.9	265	2,277	760

011110.14 Selected properties of nitrogen and oxygen at 1 atm°

T (K)	ρ (kg/m ³)	c_p (kJ/kg·K)	$\mu \cdot 10^7$ (N·s/m ²)	$\nu \cdot 10^6$ (m ² /s)	$k \cdot 10^3$ (W/m·K)	$\alpha \cdot 10^6$ (m ² /s)	Pr
Nitrogen (N₂)							
100	1.2588	1.070	68.8	2.00	9.58	2.60	0.768
150	1.2594	1.050	100.6	4.45	13.9	5.86	0.759
200	1.6555	1.043	129.2	7.65	18.3	10.4	0.756
250	1.3488	1.042	154.9	11.48	22.2	15.8	0.727
300	1.1235	1.041	178.2	15.86	25.9	22.1	0.716
350	0.9625	1.042	200.0	20.78	29.3	29.2	0.711
400	0.8425	1.045	220.4	26.16	32.7	37.1	0.704
450	0.7485	1.050	239.6	32.01	35.8	45.6	0.703
500	0.6739	1.056	257.7	38.24	38.9	54.7	0.700
550	0.6124	1.065	274.7	44.86	41.7	63.9	0.702
600	0.5615	1.075	290.8	51.79	44.6	73.9	0.701
700	0.4812	1.098	321.0	66.71	49.9	94.4	0.706
800	0.4211	1.122	349.1	82.90	54.8	116	0.715
900	0.3743	1.146	375.3	100.3	59.7	139	0.721
1000	0.3368	1.167	399.9	118.7	64.7	165	0.721
1100	0.3062	1.187	423.2	138.2	70.0	193	0.718
1200	0.2807	1.204	445.3	158.6	75.8	224	0.707
1300	0.2591	1.219	466.2	179.9	81.0	256	0.701
Oxygen (O₂)							
100	1.945	0.962	76.4	1.94	9.25	2.44	0.796
150	2.585	0.921	114.8	4.44	13.8	5.80	0.766
200	1.930	0.915	147.5	7.64	18.3	10.4	0.757
250	1.542	0.915	178.6	11.58	22.6	16.0	0.725
300	1.284	0.920	207.2	16.14	26.8	22.7	0.711
350	1.100	0.929	233.5	21.23	29.6	29.0	0.713
400	0.9620	0.942	258.2	26.84	33.0	36.4	0.737
450	0.8554	0.956	281.4	32.90	36.3	44.3	0.741
500	0.7698	0.972	303.5	39.40	41.2	53.1	0.716
550	0.6998	0.988	324.0	46.30	44.1	63.8	0.726
600	0.6414	1.003	343.7	53.59	47.3	73.5	0.729
700	0.5498	1.031	380.8	69.26	52.8	93.1	0.744
800	0.4810	1.054	415.2	86.32	58.9	116	0.743
900	0.4275	1.074	447.2	104.6	64.9	141	0.740
1000	0.3848	1.090	477.0	124.0	71.0	169	0.733
1100	0.3498	1.103	505.5	144.5	75.8	196	0.736
1200	0.3206	1.115	532.5	166.1	81.9	229	0.725
1300	0.2960	1.125	558.4	188.6	87.1	262	0.721

ภาคผนวก ข.

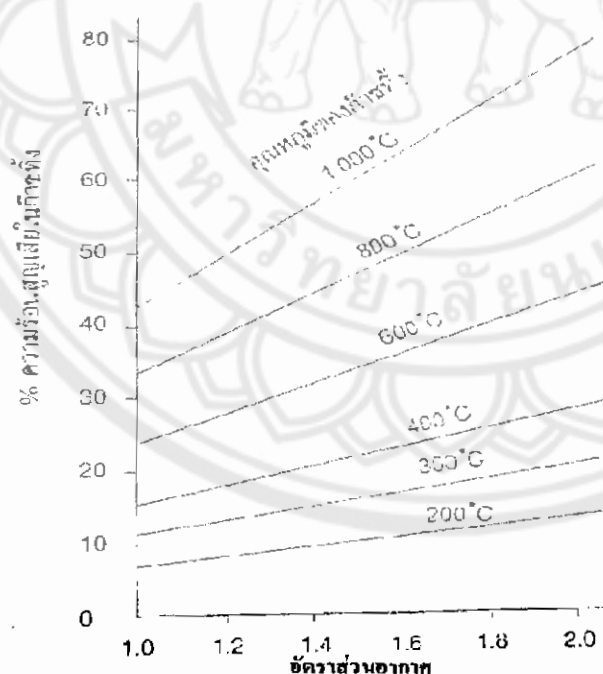
ตารางที่ ข.1 อัตราส่วนอากาศมาตรฐานสำหรับเตาอุตสาหกรรม (Industrial furnace)

กราฟที่ ข.2 ความสัมพันธ์ระหว่างอัตราส่วนอากาศกับความสูญเสียในก๊าซทิ้ง



ตาราง ข.1 อัตราส่วนอากาศมาตรฐานสำหรับเตาอุตสาหกรรม (Industrial furnace)

ประเภท	อัตราส่วนอากาศมาตรฐาน
เตาหลอมสำหรับหล่อโลหะ	1.3
เตาให้ความร้อนขึ้นเหล็กกล้าอย่างต่อเนื่อง	1.25
เตาให้ความร้อนโลหะที่ไม่ใช่เตาให้ความร้อนขึ้นเหล็กกล้าอย่างต่อเนื่อง	1.3
เตา Heat Treatment อย่างต่อเนื่อง	1.3
เตาผลิตก๊าซและเตาให้ความร้อนก๊าซ	1.4
เตาให้ความร้อนแก่น้ำมันเตา	1.4
เตา Thermal Cracking และเตาปรุงแต่งคุณภาพ	1.3
เตาเผาซีเมนต์	1.3
เตาเผาอะลูมินาและเตาเผาปูนขาว	1.4
เตาหลอมละลายแก้วอย่างต่อเนื่อง	1.3



กราฟที่ ข.2 ความสัมพันธ์ระหว่างอัตราส่วนอากาศกับความร้อนสูญเสียในก๊าซทิ้ง

ภาคผนวก ค.

ตัวอย่างการคำนวณ Adiabatic Flame Temperature โดยใช้โปรแกรม HPFLAME



**ตัวอย่าง แสดงขั้นตอนการหา Adiabatic flame temperature ที่อุณหภูมิอากาศ 298 K
โดยใช้ โปรแกรม HPFLAME**

1) ใส่ค่าต่างๆจากการคำนวณที่ผ่านมา ดังนี้

Adiabatic Flame Calculation for Specified Fuel, Phi, P, & Reactant

Enthalpy Using Olikara & Borman Equilibrium Routines

Problem Title: USER SPECIFIES TITLE HERE

1.475 /CARBON ATOMS IN FUEL

4.5 /HYDROGEN ATOMS IN FUEL

0.6 /OXYGEN ATOMS IN FUEL

0.05 /NITROGEN ATOMS IN FUEL

0.80 /EQUIVALENCE RATIO

1000 /TEMPERATURE (K) (Initial Guess)

101325.0 /PRESSURE (Pa.)

-81543.725 /ENTHALPY OF REACTANTS PER KMOL FUEL (kJ/kmol-fuel)

2) กด Save&Run จะได้

Adiabatic Flame Calculation for Specified Fuel, Phi, P, & Reactant

Enthalpy Using Olikara & Borman Equilibrium Routines

Problem Title: USER SPECIFIES TITLE HERE

Data below are as read from the input file. Compare with INPUT.HP. If they do not agree, your input data have not been entered correctly.

CARBON ATOMS	1.5	OXYGEN ATOMS	0.6
HYDROGEN ATOMS	4.5	NITROGEN ATOMS	0.1
EQUIVALENCE RATIO	0.800		
TEMPERATURE (K) guess	1000.0		
PRESSURE (Pa)	101325.0		
ENTHALPY OF REACTANTS (kJ/kmol fuel)	-81543.7		

FLAME TEMP. & COMBUSTION PRODUCTS PROPERTIES

Flame Temperature [K] = 2126.16

Mixture Enthalpy [J/kg] = -0.1908E+06

Mixture Specific Heat, Cp [J/kg-K] = 0.174287E+04

Specific Heat Ratio, Cp/Cv = 1.2106

Mixture Molecular Weight [kg/kmol] = 28.1787

Moles of Fuel per Mole of Products = 0.06593431

The mole fractions of the product species are:

H: 0.00008515 O: 0.00031365 N: 0.00000000

H2: 0.00054613 OH: 0.00282649 CO: 0.00176774

NO: 0.00442726 O2: 0.03601337 H2O: 0.14635024

CO2: 0.09548536 N2: 0.71218459 H2O: 0.14635024

3) เมื่อโปรแกรมทำการประมวลผลแล้ว จะได้ค่า Adiabatic flame temperature ที่อุณหภูมิของอากาศที่เข้าห้องเผาไหม้ 298 K รวมถึงค่าต่างๆ แต่ในที่นี้ใช้เพียงค่า Adiabatic flame temperature หลังจากนั้นสามารถหาค่า Adiabatic flame temperature ที่อุณหภูมิของอากาศใดๆที่เข้าห้องเผาไหม้ได้ โดยใส่ค่าเอนทาลปี ของสารตั้งต้นที่ได้คำนวณไว้ในตารางที่ 3.6

ภาคผนวก ง.

ตัวอย่างตารางบันทึกข้อมูลการใช้เชื้อเพลิง และก้อนอูมิเนียมที่ทำการหลอมในแต่ละวัน



MELTING DAILY REPORT

DATE 5/1/49SHIFT ☐ WHITE ☒ YELLOWDAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME									
8:00									296009
9:00									296049
10:00									296025
11:00									296085
12:00									296139
13:00									296183
14:00									296213
15:00									296270
16:00									296334
17:00									296381
18:00									296405
19:00									296447
20:00									296468
TOTAL GAS USED (M ³)									559

MELTING DAILY REPORT

DATE 6/1/49

SHIPY



WHITE



YELLOW



DAY



NIGHT



AL-DC1



AL-DC2



AL-DC3

TEMPERATURE RECORD

MATERIAL CHARGING

CHECK BY เกษมกร

FUEL CONSUMPTION REPORT									
TIME	1	2	3	4	5	6	7	8	9
20:00									297536
21:00									297599
22:00									297669
23:00									297735
00:00									297786
1:00									297820
2:00									297861
3:00									297919
4:00									297971
5:00									298004
6:00									298020
7:00									298031
8:00									298079
TOTAL GAS USED (ML)									523

MELTING DAILY REPORT

DATE 7/1/49 SHIFT ☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME							
8:00							296079
9:00							296106
10:00							296168
11:00							296232
12:00							296292
13:00							296323
14:00							296357
15:00							296416
16:00							296454
17:00							296507
18:00							296548
19:00							296559
20:00							296568
TOTAL GAS USED (L)							280

FLUX FEED

TIME START	TIME STOP	DATE	TIME	FLUX BY
12:00:00	12:00:00	12/01/2000	12:00:00	

TAMAGO CUP

800	2000
900	2100
1000	2200
1100	2300
1200	000
1300	100
1400	200
1500	300
1600	400
1700	500
1800	600
1900	700

Element		
Cu		
Cl		
Rg		
Zn		
Fe		
Na		
Ni		
Ti		
Pd		
Sn		
Cr		
Ca		

Contractor: AL - from StartLine.....To: Check By

- **Unit Stop Line.....Ton Check By**

1230. 000 00146 200 N.

- 1130-2330 CHECK BY _____

- 18.00-06.00 CHECK BY

FILTER: ☐ **NO** ☐ **YES**

REMARK

MATERIAL CHARGING

CHECK BY អ្នកប្រឹក្សា

[illegible]

MELTING DAILY REPORT

DATE 8/1/49 SHIFT ☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME							
2000							299047
2100							299054
2200							299055
2300							299077
0000							299085
0100							299100
0200							299109
0300							299119
0400							299130
0500							299142
0600							299159
0700							299167
0800							299175
TOTAL GAS USED (M ³)							128

FLUX FEED

TIME START	TIME STOP	FLUX	FLUX	FLUX	FLUX	FLUX	FLUX	FLUX BY

TARADO CUP

800	2000	
900	2100	
1000	2200	
1100	2300	
1200	0000	
1300	0100	
1400	0200	
1500	0300	
1600	0400	
1700	0500	
1800	0600	
1900	0700	

ALLOY		
Cr		
Cl		
Mg		
Zn		
Fe		
Mn		
Ni		
Ti		
Pb		
Sn		
Cu		
Ca		

Sintered AL - max Start Line.....Ton Check By.....

- max Stop Line.....Ton Check By.....

Sintered AL - max

- 1130-2330 CHECK BY.....

- 1800-0600 CHECK BY.....

FILTER BY ☐ And ☐

REMARK:

MATERIAL CHARGING

CHECK BY: 18/08/1

TIME	INGOT	RICER	CH	CH	OCAP
0600					
0615	247				
0630		252			
0640	251				
0650	257				
0940		252			
0950	278				
1000	274				
1010		252			
1020	246				
1030	254				
1100		252			
1105		252			
1125	242				
1140	245				
1150	254	63			
1400	250				
1410	243				
1420		252			
1440	243				
1450	246				
1500	244	83			
1515		252			
1525	247				
1600	246				
1635		252			
1700	245				
TOTAL	45310	10	10	133710	10
WEIGHT	566210				

MELTING DAILY REPORT

DATE 9/1/49 SHIFT ☒ WHITE ☐ YELLOW☐ DAY ☒ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE
8:00							299916
9:00							299967
10:00							300035
11:00							300095
12:00							300149
13:00							300179
14:00							300248
15:00							300307
16:00							300350
17:00							300375
18:00							300393
19:00							300425
20:00							300458
TOTAL GAS USED (M ³)							552

FLUX FEED

TIME START	TIME STOP	FLUX FEED	FLUX FEED	FLUX FEED	FLUX FEED	FLUX BY

TAPADO CUP		
8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	0:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

BENCH		
Cr		
Si		
Mg		
Ca		
Fe		
Mn		
Ti		
Pb		
Sa		
Cr		
Ca		

Checked AL - non Start Line.....Toa Check By.....

- non Stop Line.....Toa Check By.....

Checked AL - non Start Line.....Toa Check By.....

- 11:30-13:30 CHECK BY.....

- 15:00-16:00 CHECK BY.....

FILTER () ☐ 100 ☐ 1000

REMARK:

MATERIAL CHARGING

CHECK BY นายจันทิมา

TIME	INGOT	PCER	CH	CH	SCAP
20:10					35
20:25		252			
20:50	252				
21:10	254				
20:20	252				
20:35	243				
20:50	248				
22:10				155	
22:25	248				
22:40		252			
22:55	249				
23:10	252				
23:30		252			
23:50	245				
01:00	249				
01:15	245				
01:35	249				
01:55		252			
02:15	252				
02:30	247				
02:55	252				
03:15	245				
TOTAL					
WEIGHT					

MELTING DAILY REPORT

DATE 10/1/49 SHIP ✓ WHITE ☒ YELLOW ☐

☐ DAY ☒ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

MATERIAL CHARGING

CHECK BY เกรียงไกร

TIME							
20:00							301003
21:00							301053
22:00							301123
23:00							301177
0:00							301228
1:00							301319
2:00							301336
3:00							301338
4:00							301384
5:00							301432
6:00							301473
7:00							301494
8:00							301546
TOTAL GAS USED (L)							543

FLUX FEED

TIME START	TIME STOP	CONDUCTOR NAME	CONDUCTOR ID	FLUX BY

TAMAGO CUP		
8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	0:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

Element		
Cu		
Si		
Mg		
Zn		
Fe		
Hb		
Ni		
Ti		
Pb		
Sa		
Cr		
Ca		

US: INITIAL = 7500 Start Line.....To: Check By..... - 10: Stop Line.....To: Check By..... US: 1000 Filter gas H ₂ - 11:30-23:30 CHECK BY..... - 15:00-06:00 CHECK BY..... FILTER 10 ☐ 50 ☐ 100 ☐ 1500 REMARK:

[illegible]

MELTING DAILY REPORT

DATE 11/1/49 SHIFT ☐ WHITE ☒ YELLOW☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE
8:00							301546
9:00							301607
10:00							301669
11:00							301741
12:00							301799
13:00							301818
14:00							301848
15:00							301899
16:00							301967
17:00							302016
18:00							302061
19:00							302064
20:00							302082
TOTAL GAS USED (KG)							536

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	FLUX BY

TABBAG CUP

8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	00:00	
13:00	01:00	
14:00	02:00	
15:00	03:00	
16:00	04:00	
17:00	05:00	
18:00	06:00	
19:00	07:00	

8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	00:00	
13:00	01:00	
14:00	02:00	
15:00	03:00	
16:00	04:00	
17:00	05:00	
18:00	06:00	
19:00	07:00	

AL - 11:00 Start Line.....To Check By.....

- 11:00 Stop Line.....To Check By.....

AL - 11:00 Start Line.....To Check By.....

- 11:00-03:30 CHECK BY.....

- 18:00-05:00 CHECK BY.....

FILTER NO. ☐ 100 ☐ 1000

REMARK

MATERIAL CHARGING

CHECK BY: พชร

TIME	INGOT	PCER	C/H	C/H	OCAP
06:30	251				
06:50	252				
07:05				252	
07:10					50
08:20					
08:30	252				
08:40	256				
08:50		232			
09:10	248				
09:20	249				
09:30		252			
09:40	242				
09:50	245				
10:10	254	63			
10:20	250				
11:00	243				
11:10		252			
11:20	243				
11:40	246				
13:40	244	83			
13:50		252			
14:00	247				
14:10	246				
14:30		252			
14:50	245				
15:00	246				
15:20		198			
15:40	248				
15:50	247				
16:00		252			
16:10	245				
16:20	245				
TOTAL					
WEIGHT					

MELTING DAILY REPORT

DATE 11/1/49 SHIFT ☒ WHITE ☐ YELLOW☐ DAY ☒ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

MATERIAL CHARGING

CHECK BY เกษม งาม

TIME	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE
8:00							302063
9:00							302132
10:00							302196
11:00							302268
12:00							302322
13:00							302387
14:00							302405
15:00							302474
16:00							302514
17:00							302572
18:00							302604
19:00							302634
20:00							302654
TOTAL GAS USED (M ³)							571

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	FLUX

TAPAGO CUP

8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	0:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

ELEMENT		
Cu		
Si		
Mg		
Zn		
Fe		
Mn		
Ni		
Ti		
Pb		
Sn		
Cr		
Ca		

ELEMENTAL AL - 80% Start Line.....Toa Check By.....

- 10% Stop Line.....Toa Check By.....

ELEMENTAL AL - 80%

- 11:30-23:30 CHECK BY.....

- 18:00-05:00 CHECK BY.....

FILTER 0.1 ☐ 0.2 ☐ 0.5 ☐

REMARK

TIME	INGOT	RIDER	C/H	C/H	SCAP
20:30	250				
20:35	248				
20:40	246				
20:55	242				
21:20		250			
21:25	257				
21:50	249				
22:10		252			
22:15	249				
22:50	237				
23:15	241				
23:25	239				
23:50		252			
01:20	250				
01:30	249				
01:45		252			
02:00	237				
02:15	245				
02:25		252			
02:40	247				
02:50	243				
03:15		252			
03:30	245				
03:45	245				
TOTAL	10	4310 KG	10	1520 KG	10
WEIGHT	5830 KG.				

MELTING DAILY REPORT

DATE 12/1/49 SHIFT ☐ WHITE ☒ YELLOW☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE	TEMPERATURE
0900						302654
2100						302679
2200						302704
2300						302755
0000						302833
1000						302915
2000						302969
3000						302999
4000						303076
5000						303120
6000						303154
7000						303169
8000						303174
TOTAL GAS USED (KG)						500

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	FLUX BY

TANAGO CUP	
800	2000
900	2100
1000	2200
1100	2300
1200	0000
1300	1000
1400	2000
1500	3000
1600	4000
1700	5000
1800	6000
1900	7000

ITEM	ITEM	ITEM	ITEM
CH			
CI			
CU			
Fe			
Mn			
Ni			
Ti			
Pb			
Si			
Cr			
Ca			

Date and AL - mat. Start Line.....To: Check By.....

- mat. Stop Line.....To: Check By.....

Date and AL - mat. Stop Line.....To: Check By.....

- 1100-2300 CHECK BY.....

- 1800-0600 CHECK BY.....

FILTER ☐ 012 ☐ 10000

REMARK

MATERIAL CHARGING

CHECK BY: ๗๕๖๗

TIME	INGOT	POUR	CH	CH	SCAP
08:40					50
09:00	252				
09:55	251				
10:00		252			
10:05	245				
10:35	247				
10:40		252			
10:50	248				
10:55	244				
11:10		252			
11:30	254				
11:40	251				
11:50		252			
11:55	249				
12:00	237				
12:20	253				
13:25		252			
13:40	237				
13:45	254				
14:20		252			
14:25	295				
14:50	254				
15:20		145			
15:35	248				
15:50	251				
TOTAL					
WEIGHT					

MELTING DAILY REPORT

DATE 13/1/49 SHIFT ☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME							
8:00							303740
9:00							303750
10:00							303752
11:00							303850
12:00							303940
13:00							303960
14:00							304010
15:00							304045
16:00							304105
17:00							304168
18:00							304171
19:00							304207
20:00							304220
TOTAL GAS USED (M)							480

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX	FLUX
START	STOP	FLUX	FLUX	FLUX	FLUX	FLUX	FLUX

TAMAGO CUP		
8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	0:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

FLUX	FLUX	FLUX	FLUX
C4			
SI			
BJ			
Zn			
Fe			
BI			
NI			
TI			
PD			
Ca			
Cr			
Ca			

CHECK BY AL - Start Line.....To Check By.....

- 10:00 Stop Line.....To Check By.....

CHECK BY AL -

- 11:30-23:30 CHECK BY.....

- 18:00-06:00 CHECK BY.....

FILTER IN ☐ 5% ☐ 10%

REMARK:

MATERIAL CHARGING

CHECK BY 13/1/49

TIME	INDOT	REER	CH	CH	SCAP
05:00					30
05:15	247				
05:30		252			
05:40	251				
05:50	257				
05:40		252			
05:50	276				
10:00	274				
10:10		252			
10:20	246				
10:30	254				
11:00		252			
11:05	275				
11:25	255				
11:30		252			
11:50	254				
14:00	249				
14:10		252			
14:20	254				
14:40	249				
14:50		252			
15:00	258				
15:15	250				
15:25		252			
16:00	252				
16:35	250				
17:00		252			
TOTAL	2878 KG	10	2016 KG	10	10
WEIGHT	5824 KG.				

MELTING DAILY REPORT

DATE 14/1/49 SMIT
☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME							
2000							304699
2100							304720
2200							304731
2300							304747
0000							304764
0100							304781
0200							304796
0300							304814
0400							304831
0500							304845
0600							304858
0700							304877
0800							304899
TOTAL GAS USED (M)							200

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	

TAMAGO CUP

800	3000	
900	2100	
1000	2200	
1100	2300	
1200	0000	
1300	1000	
1400	2000	
1500	3000	
1600	4000	
1700	5000	
1800	6000	
1900	7000	

Si		
Al		
Mg		
Zn		
Fe		
Mn		
Ni		
Ti		
Pb		
Cu		
Cr		
Ca		

Est. mlt. AL - start Start Line.....Toa Check By.....

- mlt. Stop Line.....Toa Check By.....

Estimate mlt. m.

- 11.00-23.30 CHECK BY.....

- 18.00-06.00 CHECK BY.....

FILTER ☐ 10 ☐ 20 ☐ 30

REMARK:

MATERIAL CHARGING

CHECK BY วสุรัตน์

TIME	INGOT	REER	CH	CH	SCAP
08:30	281				
08:30		282			
09:05		282			
09:20	289				
09:35	285				
09:50		282			
10:05	251				
10:25	280				
10:55		282			
11:15	264				
11:35	258				
11:50		282			
12:45	253				
12:55	235				
13:05	279				
13:20	261				
14:05		282			
14:30	251				
14:45	254				
15:20		282			
15:50	265				
16:10	257				
TOTAL					
WEIGHT					

 3638 IG 1G 1G 176+1G 1G 1G 1G

5873 IG.

MELTING DAILY REPORT

DATE 15/1/49 SHIFT ☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DGT ☐ AL-DGZ ☒ AL-DG3

TEMPERATURE RECORD

TIME							
8:00							304699
9:00							304720
10:00							304731
11:00							304747
12:00							304764
13:00							304781
14:00							304796
15:00							304814
16:00							304831
17:00							304845
18:00							304858
19:00							304877
20:00							304899
TOTAL GAS USED (KG)							200

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	FLUX BY

TIME	TIME	FLUX
8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	0:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

FLUX	FLUX	FLUX
Cr		
Si		
Mg		
Ca		
Fe		
Mn		
Ni		
Ti		
Pb		
Cu		
Cr		
Ca		

AL - Start Line To Check By

- Stop Line To Check By

AL - Start Line To Check By

- 11:20-23:30 CHECK BY

- 18:00-05:00 CHECK BY

FILTER ☐ 100 ☐ 1000

REMARK:

MATERIAL CHARGING

CHECK BY วสุรัตน์

TIME	INGOT	REER	CH	CH	ODAP
01:10	247				
01:20	248				
01:30		252			
06:50	242				
09:05	252				
09:15		252			
09:30	252				
09:35	244				
09:55		252			
10:15	249				
10:20	248				
10:35	245				
10:45	245				
11:05		252			
11:20	243				
11:40		180			
13:40	248				
13:45	245				
14:00		252			
14:05	247				
14:15	245				
14:55		252			
15:20	245				
15:30	244				
15:50		198			
17:30	249				
17:35	249				
TOTAL	10 5182 KG	10 2109 KG	10	10	10
WEIGHT	7542 KG.				

MELTING DAILY REPORT

DATE 17/1/49 SHIFT ☒ WHITE ☐ YELLOW☐ DAY ☒ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME							
8:00							307281
9:00							307322
10:00							307396
11:00							307453
12:00							307482
13:00							307556
14:00							307589
15:00							307646
16:00							307704
17:00							307759
18:00							307777
19:00							307851
20:00							307901
TOTAL GAS USED (B)							620

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	FLUX BY

TIME	TIME	FLUX
8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	00:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

FLUX	FLUX	FLUX
Ca		
Si		
Mn		
Fe		
Mn		
Fe		
Pb		
Si		
Cr		
Ca		

Start Line - non Start Line.....To Check By.....

- non Stop Line.....To Check By.....

Check Line Use N.

- 11:30-23:30 CHECK BY.....

- 18:00-06:00 CHECK BY.....

FILTER ☐ ☐ ☐ ☐

REMARK

MATERIAL CHARGING

CHECK BY เกรียงไกร

TIME	INGOT	RICER	C/H	C/H	SCAP
20:10	216				
20:20	257				
20:30		252			
20:50	252				
21:30	252				
21:40		252			
21:50	257				
22:30		252			
23:00	247				
23:30		252			
23:50	258				
23:55	250				
01:25		252			
01:35	254				
01:50	254				
02:50		252			
03:15	257				
03:40	251				
03:50		252			
04:35	257				
04:50	253				
05:00		252			
05:40	259				
06:15	254				
06:50		252			
07:10	253				
07:30	270				
07:40		255			
07:45	259				
07:55	261				
TOTAL	5111 IG	10	10	2963 IG	10
WEIGHT	7674 IG.				

MELTING DAILY REPORT

DATE 18/1/49

☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

MATERIAL CHARGING

CHECK BY วสุรัตน์

TIME							
20:00							307901
21:00							307972
22:00							308037
23:00							308090
0:00							308180
1:00							308187
2:00							308233
3:00							308301
4:00							308373
5:00							308408
6:00							308452
7:00							308476
8:00							308520
TOTAL GAS USED (M ³)							619

FLUX FEED

TIME START	TIME STOP	CONDUCT	CHLORIDE	WATER	PH	FLUX BY
		mg/dm ³	mg/dm ³	mg/dm ³		

YAMAGO CUP		
600	2000	
900	2100	
1000	2200	
1100	2300	
1200	000	
1300	100	
1400	200	
1500	300	
1600	400	
1700	500	
1800	600	
1900	700	

Al		
Br		
Cu		
Si		
Mg		
Zn		
Fe		
Mn		
Ni		
Ti		
Pb		
Sn		
Cr		
Ca		

```

          LALINE AL - 000 StartLine.....To: Check By.....
          - 000 Stop Line.....To: Check By.....

LALINE AL - 000 M,
- 11.30-23.30 CHECK BY .....
- 18.00-06.00 CHECK BY .....

FILTER 10 ☐ 0 12 ☐ 12.00

```

REMARK

[illegible]

MELTING DAILY REPORT

DATE 18/1/49 SHIFT ☒ WHITE ☐ YELLOW

☐ DAY ☒ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

MATERIAL CHARGING

CHECK BY เกรียงไกร

TIME							
20:00							308520
21:00							308506
22:00							308552
23:00							308724
00:00							308754
1:00							308781
2:00							308831
3:00							308896
4:00							308908
5:00							308975
6:00							308968
7:00							308993
8:00							308008
TOTAL GAS USED (M ³)							488

FLUX FEED

TIME START	TIME STOP	WIND DIRECTION	WIND SPEED	WIND GUST	WIND GUST	FLUX BY
		0-10000	200-2500	200-2500	200-2500	

TAYAGO CUP	
800	2000
900	2100
1000	2200
1100	2300
1200	000
1300	100
1400	200
1500	300
1600	400
1700	500
1800	600
1900	700

Element		
Cl		
Si		
B		
Zn		
Fe		
Mn		
Ni		
Ti		
Pd		
Sn		
Cr		
Ca		

150.0000	AL -	Start Line.....	To: Check By.....
- 1000 Stop Line.....To: Check By.....			
Use start to end card H.			
- 1130-0330	CHECK BY		
- 1500-0600	CHECK BY		
FILTER	is	☐ on	☐ clean.
REMARK:			

[illegible]

MELTING DAILY REPORT

DATE: 19/1/49 SHIFT ☐ WHITE ☒ YELLOW

☒ DAY ☐ NIGHT ☐ AL-DC1 ☐ AL-DC2 ☒ AL-DC3

TEMPERATURE RECORD

TIME							
8:00							309008
9:00							309060
10:00							309140
11:00							309200
12:00							309274
13:00							309314
14:00							309363
15:00							309415
16:00							309473
17:00							309491
18:00							309508
19:00							309555
20:00							309594
TOTAL GAS USED (M ³)							586

FLUX FEED

TIME	TIME	FLUX	FLUX	FLUX	FLUX	FLUX BY
START	STOP	FLUX	FLUX	FLUX	FLUX	

TABAGO CUP

8:00	20:00	
9:00	21:00	
10:00	22:00	
11:00	23:00	
12:00	0:00	
13:00	1:00	
14:00	2:00	
15:00	3:00	
16:00	4:00	
17:00	5:00	
18:00	6:00	
19:00	7:00	

BR		
CU		
GI		
SI		
DI		
FE		
BA		
NI		
TI		
Pb		
Ca		
Cr		
Ca		

50.000 AL - min Start Line.....For Check By.....

- min Stop Line.....For Check By.....

50.000 AL - min Start Line.....For Check By.....

- min Stop Line.....For Check By.....

- min Stop Line.....For Check By.....

FILTER in: ☐ 5N ☐ 0.5mm

REMARK:

MATERIAL CHARGING

CHECK BY: 25/1/49

TIME	INGOT	RCR	CH	C/H	SCAP
08:00					63
08:10	245				
08:25	254				
08:45		252			
09:00	235				
09:15	265				
09:30		252			
09:45	237				
10:00	283				
10:15		252			
10:30	270				
10:50	252				
11:05		252			
11:25	253				
11:40	254				
12:00		252			
13:25	254				
13:40	260				
13:55		252			
14:20	253				
14:40	247				
14:55		252			
15:30	237				
TOTAL	3851 KG	10	10	1764 KG	10
WEIGHT	5678 KG				

ภาคผนวก จ.

ลำดับสัญลักษณ์

Nomenclature

1J	=	2.3885×10^{-4} kcal	=	Energy
3600s	=	1 hr	=	Time
a		Molar oxygen-fuel (kmol/kmol)		
A/F		Mass air-fuel ratio (kg/kg)		
$c_p, \bar{c}_p$		Constant-pressure specific heat (J/kg-K or J/kmol-K)		
$c_v, \bar{c}_v$		Constant-volume specific heat (J/kg-K or J/kmol-K)		
F/A		Mass fuel-air ratio (kg/kg)		
$h_f^\circ, \bar{h}_f^\circ$		Enthalpy of formation (J/kg or J/kmol)		
$H, h, \bar{h}$		Enthalpy (J or J/kg or J/kmol)		
$\Delta H_c, \Delta h_c, \Delta \bar{h}_c$		Heat of combustion (Heating value) (J or J/kg or J/kmol)		
$\Delta H_R, \Delta h_R, \Delta \bar{h}_R$		Enthalpy of reaction (J or J/kg or J/kmol) =		
HHV		Higher heating value (J/kg)		
LHV		Lower heating value (J/kg)		
m		Mass (kg)		
$\dot{m}$		Mass flow rate (kg/s)		
MW		Molecular weight (kg/kmol)		
N		Number of moles (kmol)		
P		Pressure (Pa)		
Q, q		Heat (J or kJ)		
$\dot{Q}, \dot{q}$		Heat transfer rate (J/s = W or W/kg)		
R		Specific gas constant (J/kg-K)		
R_u		Universal gas constant (J/kmol-K)		
T		Temperature (K)		
v		Velocity (m/s)		
V, v		Volume (m ³ or m ³ /kg)		
x		Number of carbon atom in fuel		

y	Number of hydrogen atom in fuel
Y	Mass fraction (kg/kg)

Greek Symbols

Φ	Equivalence ratio
ρ	Density (kg/m^3)
χ	Mole fraction

Subscripts

ad	Adiabatic
A	Air
cv	Control volume
f	Final of formation
F	Fuel
g	Gas
i	i th species or inlet
init	Initial
l	Liquid
mix	Mixture
o	Outlet
prod	Product
reac	Reactant
ref	Reference
s	Sensible
sat	Saturation state
stoic	Stoichiometric
T	At temperature T